Quinazoline Antifolate Thymidylate Synthase Inhibitors: Benzoyl Ring Modifications in the C2-Methyl Series

Peter R. Marsham,^{*,†} Ann L. Jackman,[‡] John Oldfield,[†] Leslie R. Hughes,[†] Timothy J. Thornton,[‡] Graham M. F. Bisset,[‡] Brigid M. O'Connor,[‡] Joel A. M. Bishop,[‡] and A. Hilary Calvert^{‡,§}

ICI Pharmaceuticals, Mereside, Alderley Park, Macclesfield, Cheshire SK10 4TG, England, and Institute of Cancer Research, 15 Cotswold Road, Sutton, Surrey SM2 5NG, England. Received March 13, 1990

The synthesis of nine new 2-methyl-10-propargylquinazoline antifolates with substituents in the *p*-aminobenzoyl ring is described. In general the synthetic route involved the coupling of the appropriate ring-substituted diethyl N-[4-(prop-2-ynylamino)benzoyl]-L-glutamate with 6-(bromomethyl)-3,4-dihydro-2-methyl-4-oxoquinazoline (11) followed by deprotection using mild alkali. The compounds were tested as inhibitors of partially purified L1210 thymidylate synthase (TS). They were also examined for their inhibition of the growth L1210 cells in culture. Compared to the parent compound 1a the 2'-fluoro analogue 2a exhibited enhanced potency in both systems whereas the 3'-fluoro analogue 3a showed enhanced growth inhibitory properties against L1210 cells despite being a poorer inhibitor of the isolated enzyme. Chloro, hydroxy, methoxy, and nitro substituents in the 2'-position were also well tolerated by the enzyme but failed to give enhanced growth inhibition. The series was extended to cover analogues of the 2'-fluoro, 3'-fluoro, 2'-chloro, 2'-methyl, 2'-amino, 2'-methoxy, and 2'-nitro derivatives with modified alkyl substituents at N10.

Introduction

In the preceding article¹ we have demonstrated that the introduction of a fluorine atom into the 2'-position of a series of quinazoline antifolates enhances their potency as inhibitors of both thymidylate synthase (TS) and cell growth. This indicated that modification of the benzoyl ring could offer opportunities for further modulation of the biological activities of these compounds. In the work described below we have sought to optimize the stereoe-lectronic requirements of this region of the molecule by exploring, in the highly potent C2-methyl series,^{2,3} a wider range of ring substituents than hitherto studied.

Chemistry

The majority of the compounds (2-9) were prepared by the route outlined in Scheme I and involved the condensation of the (bromomethyl)quinazolinone 11^3 with the appropriate N-alkylated anilines 36-43 using either CaCO₃ (method A) or 2,6-lutidine (method P) to scavenge HBr.



The resulting antifolate diesters 44-51 were hydrolyzed by aqueous alkali (method B) to yield the antifolate diacids listed in Table II. The N-alkylated anilines 36a,¹ 38a,⁴ and $40a^4$ were known compounds. The others were prepared (Scheme I) from the appropriate nitrobenzoic acids 12-19. The derived acid chlorides were condensed with diethyl L-glutamate (methods K and O), and the resulting (nitrobenzoyl)-L-glutamate diesters 20-27 were reduced (H₂, 10% Pd-C; method C) to the anilines 28-35. The simple N-alkylated anilines 36-43 were derived from the parent anilines by reaction with the appropriate alkyl halide under a variety of conditions according to the reactivity of the halide (methods D, E, L, M; Table I). The precursor



^a See Table I for definition of R. ^b(a) $(COCl)_2$ or $SOCl_2$; (b) diethyl glutamate, Et₃N; (c) H₂, 10% palladium on carbon; (d) RX; (e) CaCO₃, DMA or 2,6-lutidine, DMF; (f) 1 N aqueous NaOH, EtOH.

aniline **36g** (method F) for the 2'-fluoro-10-(hydroxyethyl)analogue **2g** was prepared with the hydoxyl group

[†]ICI Pharmaceuticals.

[‡]Institute of Cancer Research.

[§]Present address: Cancer Research Unit, Medical School, Framlington Place, Newcastle upon Tyne NE2 4HH, England.

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Table I. Preparation of Substituted (Aminobenzoyl)glutamate Diesters 36-43, 53, and 55 and Derived Antifolate Diesters44-51, 56, and 62

	starting	method	%	derived	%			
compd	amíne	(temp, °C)	yield	antiofolate diester	method	yield	Х	R
36aª	28	L(120) ^a	60	44 a	Α	37	2'-F	CH ₂ C=CH
28		C(20)	99	44b	Α	65	2' -F	H
36c	28	D(60)	31	44c	Α	70	2' -F	CH_3
36d	28	D(100)	52	44d	Α	62	2' -F	CH_2CH_3
36e	28	D(100)	51	44 e	Α	28	2' -F	$CH_2CH=CH_2$
36f	28	E(120)	83	44f	Α	38	2' -F	$(CH_2)_2F$
36g	28	F(140)	47	44 g	Α	2 9	2' -F	$(CH_2)_2OCOC_3H_7$
36h	28	G(80)	31	44h	Α	70	2′-F	(CH ₂) ₃ OCOCH ₃
53 ^b	52 ^b	H(130)	3 9	54^{b}	Α	27	2'-F	$(CH_2)_2$ NPhth
55 ⁶	52 ^b	F(85)	91	56 ⁶	Α	20	2′-F	CH ₂ CN
36k	28	F(110) ^c	41	44k	Α	38	2' -F	CH_2CN
37a	29	L(120)	53	45a	\mathbf{A}^{d}	40	3' -F	CH₂C≡CH
29		C(20)	99	45b	Α	40	3' -F	Н
37c	29	D(100)	21	45c	Α	81	3' -F	CH ₃
37d	29	D(100)	44	45d	Α	48	3' -F	CH_2CH_3
37e	29	D(100) ^e	36	45e ^f	N	50	3' -F	$CH_2CH=CH_2$
37f	29	E(120)	78	45f	Р	23	3' -F	$(CH_2)_2F$
37g	29	D(100)	26	45g ⁷	N	42	3' -F	$(CH_2)_2OCOCH_3$
37h	29	M(150)	27	45h [/]	N	56	3' -F	$(CH_2)_3OCOCH_3$
38a″	30 ^e	g	41	46a	Р	23	2'-Cl	$CH_2C = CH$
38d	30	M(85)	25	46d	P	38	2'-Cl	CH_2CH_3
39a	31	L(100)	65	47a	\mathbf{P}^{h}	64	2'-CF ₃	$CH_2C = CH$
40a#	32^{g}	g	66	48a	Р	35	2′-CH ₃	CH ₂ C=CH
40c	32^{g}	M(40)	22	48c	Р	47	$2'$ -CH $_3$	CH_3
40d	32^g	M(40)	68	48 d	Р	58	$2'$ -CH $_3$	CH₂CH₃
41 a	33	M(60)	62	49 a	Р	44	2'-NHCOCF ₃	CH₂C≡CH
41c	33	M(60)	44	49c	Р	63	2'-NHCOCF ₃	CH_3
41 d	33	M(60)	54	49 d	Р	57	2′-NHCOCF ₃	CH_2CH_3
42a	34	M(80)	61	50a	Р	71	2'-0C0CH ₃	CH₂C≡CH
43 a	35	M(80)	58	51 a	Р	35	2'-OCH ₃	CH ₂ C≡CH
43d	35	M(80)	48	51d	Р	54	2'-OCH ₃	CH_2CH_3
				62a	Q	73	2′-NO ₂	$CH_2C = CH$
				62d	Q	41	$2-NO_2$	CH_2CH_3

^aSee also ref 1. ^bDi-*tert*-butyl esters. ^cK₂CO₃ was used as the base. ^dProton Sponge was used as the base. ^e2,6-Lutidine was used as the base. ^fProtected as the 3-(pivaloyloxy)methyl derivative. ^gSee ref 4. ^hMgO was used as the base.

Scheme II^a



3 h

 $^{\rm a}$ (a) 2,6-Lutidine, 105 °C (method N); (b) 1 N aqueous NaOH, EtOH (method B).

protected as the butyrate ester because the corresponding acetate ester was inseparable by chromatography from the starting aniline 28. For a similar reason, 36h was prepared in two steps (method G) by alkylation of 28 with 1,3-dibromopropane followed by NaOAc displacement of the second bromide. For the synthesis of the 2'-fluoro-10-(aminoethyl) analogue 2i the aniline di-*tert*-butyl ester 52^5 was alkylated with (2-bromoethyl)phthalimide to afford 53 (method H). This was coupled to 11 in the usual way and the resulting phthalimido di-tert-butyl ester 54 deprotected to 2i in two stages by treatment with 3-(dimethylamino)propylamine followed by aqueous alkali at 60 °C (method I). The 2'-fluoro-10-cyanomethyl antifolate 2j was also prepared via the di-tert-butyl ester 52. In this case, N-cyanomethylation was achieved by reaction with bromoacetonitrile in the presence of 2,6-lutidine and NaI. The intermediate N-cyanomethylaniline 55 was coupled with 11 to give 56, which was deprotected by treatment with CF_3CO_2H to yield 2j (method J). When this sequence was repeated on the diethyl ester 28 and the N-cyanomethyl diester 44k hydrolyzed with aqueous NaOH, complete hydration of the nitrile was observed, yielding the N-carbamoylmethyl antifolate 2k. Some of the 3'-fluoro antifolates were prepared by the same methods as their 2'-fluoro counterparts, but in general the 3'-fluoro-N-alkylanilines 37a-h were relatively poor nucleophiles, ne-

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cessitating modified coupling procedures in some cases. In the coupling of **37a** to 11 the highest yield (40%) of **45a** was obtained when Proton Sponge⁶ was used as base. If the coupling of 11 to anilines in the 3'-fluoro series was performed at elevated temperatures (>90 °C) in an attempt to accelerate the reaction, alkylation of the product on N3 by a second molecule of 11 became a serious problem. We circumvented this problem by protecting 11 as its N3-(pivaloyloxy)methyl derivative **57**.⁷ This protected bromomethyl compound reacted cleanly with anilines at elevated at elevated temperatures (~105 °C) to give satisfactory yields of (pivaloyloxy)methyl-protected antifolate diesters in the 3'-fluoro series. A typical procedure of this type is outlined in Scheme II (method N).

In the synthesis of the 2'-amino (7a,c,d) and 2'-hydroxy (8a) analogues the functionalities were protected as the trifluoroacetyl and acetyl derivatives, respectively. These protecting groups were carried intact through the whole sequence and removed during the final hydrolysis with aqueous alkali. The synthesis of the 2'-nitro compound 10a required a fundamentally different order of assembly of the synthons (Scheme III). Methyl 4-amino-2-nitrobenzoate (58)⁸ was sequentially propargylated and alkylated with 11 to yield the tertiary amino ester 60 in 23%overall yield. This was saponified and the derived carboxylic acid 61 was coupled via the in situ generated azide with diethyl L-glutamate. The resulting antifolate diester 62a was hydrolyzed according to method B to yield 10a The analogous sequence in which 58 was (Table II). alkylated with iodoethane afforded 10d.

Results and Discussion

The IC₅₀ values for the inhibition of partially purified L1210 TS and for growth inhibition of L1210 cells were obtained as described in the preceding article¹ and are shown in Table II. The 2'-fluoro analogue 2a showed a 2-fold enhancement over the parent 1a in potency against TS and a 3-fold enhancement against L1210 cells. On the other hand, 3'-fluoro substitution (3a) was highly detrimental to TS inhibition. Nevertheless this 3'-fluoro derivative was still a more potent growth inhibitory agent than 1a. The fact that thymidine alone prevents the growth-inhibitory properties of 1a, 2a, and 3a suggests that



^a (a) Propargyl bromide; (b) bromomethyl compound 11, 2,6lutidine, DMF; (c) 1 N aqueous NaOH, EtOH; (d) diethyl glutamate, diphenyl phosphorazidate, Et_3N , DMF.

TS is their sole locus of action of action. A possible explanation for this apparent discrepancy is that the 3-fluoro derivative is more readily transported into cells and/or metabolized intracellularly to polyglutamate derivatives that are not only retained within the cells but also have much greater TS inhibitory potency than the parent monoglutamates. In support of this hypothesis it has been demonstrated that these fluorinated compounds have retained or improved activity as substrates for folylpolyglutamate synthetase (FPGS).⁹ We have reported¹ that there is a spin-spin coupling (J = 6.4 Hz) observed between the 2'-fluorine atom and the amide proton of 2a, suggesting that there is a hydrogen bond between these two atoms that holds this region of the molecule in a favorable conformation for binding to TS. The sharp drop in affinity for TS caused by 3'-fluoro substitution is consistent with the resulting increase in electron density in the C3' region giving rise to a direct unfavorable interaction with the enzyme or to effects on the conformation of the C9,N10 bridge region. Because of the enhanced inhibition of cell growth resulting from fluorine substitution we have replaced the N10-propargyl group of 2a and 3a with a series of alternative N10-substituents (2b-k and 3b-h). In general, 2'-fluoro substitution resulted in improved TS and L1210 cell growth inhibition across the range of N10-substituents studied (for parent compound IC_{50} values,³ see Table II). The enhancements were particularly marked (\sim 5-fold) in N10-ethyl (2d), N10-fluoroethyl (2f), N10-hydroxyethyl (2g), and N10-alkyl (2e), whereas the N10-methyl (2c) analogue showed the greatest enhancement (6-fold) in growth inhibition. Three N10-substituents, namely cyanomethyl (2j), carbamoylmethyl (2k), and aminoethyl (2i) have been prepared only in the 2'-fluoro

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Table II.	Preparation	and in V	Vitro .	Activities	of G	uinazoline	Antifolate	Diacids	2-10
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							mass		inhibn of L1210	L1210 cell growth in
							spectra	inhibn	cell growth	the presence
				%			m/z	of TS:	in culture:	of thymidine
compd	Х	R	method	yield	mp, °C	formula ^a	[M – H] ⁻	$IC_{50}, \mu M$	IC ₅₀ , μM	(% control)
la ^b	Н	CH ₂ C=CH			165	$C_{25}H_{24}N_4O_6 \cdot 2H_2O$		0.040	0.090	100
1 b ^b	Н	Η			197-201	$C_{22}H_{22}N_4O_6 \cdot H_2O$		4.50	0.07	
1c ^b	н	CH₃			254-257	$C_{23}H_{24}N_4O_60.75H_2O$		0.30	0.11	100
$1d^b$	Н	CH_2CH_3			221-225	$C_{24}H_{26}N_4O_6 \cdot 0.5H_2O$		0.17	0.36	88
1e ^b	Н	CH ₂ CH-CH ₂			188	$C_{25}H_{26}N_4O_6 \cdot 1.5H_2O$		0.48	0.17	104
$1f^{b}$	Н	$(CH_2)_2F$			207-210	$C_{24}H_{25}FN_4O_6 \cdot 1.25H_2O$		0.24	0.4	
$1\mathbf{g}^{b}$	н	$(CH_2)_2OH$			>300	$C_{24}H_{26}N_4O_7 \cdot 1.5H_2O$		0.50	0.24	
1h ^b	Н	(CH ₂) ₃ OH			300	$C_{25}H_{28}N_4O_7H_2O$		0.54	1.24	9 5
2a	2′-F	$CH_2C = CH$	в	77	228-230	$C_{25}H_{23}FN_4O_6$	493	0.020	0.027	97
2b	2′•F	Н	в	90	190–194	$C_{22}H_{21}FN_4O_6H_2O$	455	3.78	0.064	94
2c	2′-F	CH_3	в	83	224-226	$C_{23}H_{23}FN_4O_6H_2O$	469	0.12	0.019	98
2d	2′-F	CH_2CH_3	в	94	214-217	$C_{24}H_{25}FN_4O_{6}0.75H_2O$	483	0.045	0.065	96
2e	2′-F	$CH_2CH=CH_2$	в	84	20 9 –211	$C_{25}H_{25}FN_4O_6$	495	0.076	0.10	97
2f	2′-F	$(CH_2)_2F$	в	63	220-225	$C_{24}H_{24}F_2N_4O_6.0.7H_2O$	501	0.043	0.13	98
2g	2′-F	$(CH_2)_2OH$	в	90	190–196	$C_{24}H_{25}FN_4O_7H_2O$	499	0. 0 78	0.07	
2 h	2′-F	(CH ₂) ₃ OH	в	77	132-136	$C_{25}H_{27}FN_4O_7H_2O$	513	0.59	0.030	9 5
2i	2′-F	$(CH_2)_2NH_2$	I	20	210–215 ^c	$C_{24}H_{26}FN_5O_6 \cdot 1.25H_2O^d$	498	11.72	2.7	
2j	2′-F	CH_2CN	J	84	$160 - 162^{\circ}$	$C_{24}H_{22}FN_5O_6.0.5H_2O'$	494	0.17	0.034	90
2k	2′-F	CH_2CONH_2	B ^g	22	185 ^h	$C_{24}H_{24}FN_5O_7H_2O$	512	1.34	0.80	
3a	3′-F	$CH_2C \equiv CH$	в	79	156-160	$C_{25}H_{23}FN_4O_6H_2O$	493	1.43	0.052	102
3b	3′-F	Н	в	92	220-222	$C_{22}H_{21}FN_4O_6H_2O$	455	9.34	0.12	86
3c	3' -F	CH_3	в	94	210-211	$C_{23}H_{23}FN_4O_6\cdot H_2O$	469	1.14	0.14	92
3d	3′-F	CH_2CH_3	В	86	167-168	$C_{24}H_{25}FN_4O_6 \cdot 1.5H_2O$	483	0.68	0.086	96
3e	3′∙F	$CH_2CH=CH_2$	\mathbf{B}^{i}	63	108-110	$C_{25}H_{25}FN_4O_6\cdot 2H_2O$	495	5.16	0.14	97
3f	$3' \cdot \mathbf{F}$	$(CH_2)_2F$	В	83	162-166	$C_{24}H_{24}F_2N_4O_6\cdot 1.5H_2O$	501	1.75	0.069	
3g	3'-F	$(CH_2)_2OH$	B	28	120 - 122	$C_{24}H_{25}FN_4O_7 \cdot 2H_2O$	499	3.58	0.085	95
3h	3′•F	$(CH_2)_3OH$	B,	70	127 - 132	$C_{25}H_{27}FN_4O_7\cdot 2H_2O$	513	0.98	0.17	
4a	2'.CI	$CH_2C \equiv CH$	В	66	162-164	$C_{25}H_{23}CIN_4O_6 \cdot 0.5H_2O$	509	0.074	0.25	
4d	2'-C1	CH_2CH_3	В	76	201-207	$C_{24}H_{25}CIN_4O_6\cdot H_2O$	499	0.12	0.87	
5a	2'-CF ₃	$CH_2C \equiv CH$	В	85	197-198	$C_{26}H_{23}F_3N_4O_6\cdot 1.25H_2O$	543	0.48	1.2	
6a	2'-CH ₃	$CH_2C \equiv CH$	В	74	95 ⁷	$C_{26}H_{26}N_4O_6\cdot H_2O$	489	0.50	0.36	97
6c	$2' \cdot CH_3$	CH ₃	В	62	219-221*	$C_{24}H_{26}N_4O_6\cdot 2H_2O'$	465	0.94	0.8	
6d	$2'-CH_3$	CH ₂ CH ₃	В	67	212-215*	$C_{25}H_{28}N_4O_6 \cdot 0.5H_2O$	479	0.32	1.2	94
7a	$2' \cdot NH_2$	$CH_2C \equiv CH$	\mathbf{B}^m	88	182-186	$C_{25}H_{25}N_5O_6.0.5H_2O$	490	0.29	0.36	
7c	2′-NH2	CH ₃	B ^m	86	188-190*	$C_{23}H_{25}N_5O_6 \cdot 1.5H_2O$	466	0.86	0.76	
7d	$2'-NH_2$	CH_2CH_3	Bm	91	170-175*	$C_{24}H_{27}N_5O_6H_2O$	480	0.31	1.3	
8a	2'-OH	CH ₂ C=CH	Bm	83	210*	$C_{25}H_{24}N_4O_7 \cdot 0.75H_2O$	491	0.09	0.5	77
9a	2'-OCH ₃	CH₂C≡CH	B	50	175 - 180	$C_{26}H_{26}N_4O_7 \cdot 2.5H_2O$	505	0.067	3.0	95
9 d	2′-OCH ₃	CH ₂ CH ₃	B	64	152 - 157	$C_{25}H_{28}N_4O_7H_2O$	495	0.19	18.0	
10a	2′-NO ₂	CH₂C≡CH	В	73	190-202	$C_{25}H_{23}N_5O_8H_2O$	520	0.13	0.40	
10 d	2'-NO ₂	CH ₂ CH ₃	В	68	192-200*	$C_{24}H_{25}N_5O_8 \cdot 1.4H_2O^n$	510	0.145	1.5	

^aAnal. C, H, N except where stated otherwise. ^bSee ref 3. ^cSoftens >190 °C. ^dN: calcd, 13.4; found, 12.7. ^eSoftens >145 °C. [/]N: calcd, 13.9; found, 13.3. ^ePrepared from the N10-cyanomethyl ester 44k. ^hSoftens >170° C. ⁱThe starting quinazoline antifolate diethyl ester was protected as the N3-(pivaloyloxy)methyl derivative. ^jSinters above this temperature but does not give a discrete melting point. ^kDecomposes at this temperature. ⁱH: calcd, 5.9; found, 5.4. ^mHydrolyses of 49a,c,d and 50a were performed at 50 °C for 2 h. ⁿH: calcd, 5.2; found, 4.6.

series. A cyanomethyl group is a good isosteric replacement for propargyl since 2j exhibits high potency against both parameters although the bulkier carbamoylmethyl group in 2k causes diminished activity. The reduced potency of the aminoethyl (2i) analogue may result from repulsive electronic interactions with basic groups in this region of the enzyme. Steric factors alone will not explain this poor activity since the close hydroxyethyl isostere 2g is a highly potent compound. In the range of N10-substituents studied 3'-fluoro substitution (3a-h) caused a drop in TS inhibitory potency although this effect is more pronounced in the propargyl analogue than in the others. The loss of enzyme inhibition seen with these analogues was not paralleled by a similar loss in L1210 growth inhibition. Indeed a notable improvement was found with aliphatic substituents other than methyl.

Of the 2'-substituents larger than fluorine that were incorporated some, chlorine (4a,d), hydroxide (8a), methoxide (9a,d), and the nitro group (10a,d) were well tolerated by TS while others, the amino (7a,c,d), methyl (6a,c,d), and trifluoromethyl (5a) groups, were less well accommodated. However all of these substituents caused diminished inhibition of L1210 cell growth.

Overall these results demonstrate that within this series of antifolates the simultaneous variation of the substituents on N10, C2', and C3' has separate effects on the inhibitory potencies against TS and cell growth. In particular the combination of 2'-fluoro or 3'-fluoro substitution with appropriate N10-substituents has given the most potent cytotoxic TS inhibitors so far reported. The lack of direct correlation between potency against TS and the observed inhibition of cell growth suggests that other factors such as rate of uptake into the L1210 cells and degree of polyglutamation are also strongly influenced by modifications to these regions of the molecule.

Experimental Section

The general procedures used were described in the earlier $paper^{10}$ in this series.

Diethyl N-[4-[N-[(3,4-Dihydro-2-methyl-4-oxo-6quinazolinyl)methyl]-N-prop-2-ynylamino]-2-fluorobenzoyl]-L-glutamate (44a). Method A. A mixture of 36a¹ (1.10 g, 2.91 mmol), the bromomethyl compound 11³ (736 mg, 2.91 mmol), and powdered CaCO₃ (583 mg, 5.83 mmol) in DMA (9 mL) was stirred for 24 h at 50 °C under argon. The cooled mixture was filtered and the filtrate was evaporated to dryness. The crude product was purified by chromatography using a gradient of 2-6%

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v/v EtOH in CH₂Cl₂ as eluent to give an amorphous solid: 589 mg (37%); mp 202-204 °C; NMR (Me₂SO-d₆) δ 1.2 (t, 6 H, 2 OCH₂CH₃), 2.0 (m, 2 H, CHCH₂CH₂CO₂Et), 2.35 (s, 3 H, CH₃), 2.4 (t, 2 H, CHCH₂CH₂CO₂Et), 3.25 (t, 1 H, C=CH), 4.05, 4.1 (2 q, 4 H, 2 OCH₂CH₃), 4.35 (br s, 2 H, CH₂C=C), 4.4 (m, 1 H, CH), 4.8 (br s, 2 H, ArCH₂N<), 6.65 (dd, 1 H, 3'-H), 6.7 (dd, 1 H, 5'-H), 7.5 (t, 1 H, 6'-H), 7.55 (d, 1 H, quinazoline 8-H), 7.7 (dd, 1 H, quinazoline 7-H), 7.95 (d, 1 H, quinazoline 5-H), 8.15 (dd, 1 H, CONH). Anal. (C₂₉H₃₁FN₄O₆) C, H, N, F.

The procedure was repeated with the appropriate amines to yield the antifolate diesters 44b-45d (Table I).

N-[4-[N-[(3,4-Dihydro-2-methyl-4-oxo-6-quinazolinyl)methyl]-N-prop-2-ynylamino]-2-fluorobenzoyl]-L-glutamic Acid (2a). Method B. The diester 44a (360 mg, 0.65 mmol) was stirred for 5 h under argon in a mixture of 1 N aqueous NaOH (2.2 mL, 2.2 mmol), EtOH (12 mL), and H₂O (12 mL). The resulting solution was evaporated below 30 °C to ca. 10 mL, filtered into a centrifuge tube, and brought a pH 3.0 with 2 N aqueous HCl. The precipitate was isolated by centrifugation and freed from inorganic ions by repeated cycles of aqueous suspension-centrifugation-decantation until the supernatant was free of chloride ion (AgNO₃ test). The damp product was vacuum dried to give an amorphous white solid: 249 mg (77%); mp 228-230 °C; NMR (Me_2SO-d_6) δ 2.0 (m, 2 H, CHC $H_2CH_2CO_2H$), 2.35 (t, 2 H, CHCH₂C H_2 CO₂H), 2.35 (s, 3 H, CH₃), 3.25 (t, 1 H, C=CH), 4.35 (d, 2 H, CH₂C=C), 4.35 (m, 1 H, CH), 4.8 (br s, 2 H, ArCH₂N<), 6.65 (dd, 1 H, 3'-H), 6.7 (dd, 1 H, 5'-H), 7.55 (t, 1 H, 6'-H), 7.55 (d, 1 H, quinazoline 8-H), 7.7 (dd, 1 H, quinazoline 7-H), 7.95 (d, 1 H, quinazoline 5-H), 7.95 (dd, 1 H, CONH); MS (FAB) m/z 493 [M - H]⁻. Anal. (C₂₅H₂₃FN₄O₆) C, H, N.

The procedure was repeated with the appropriate diethyl esters 44b-51d and 62a,d to yield the antifolates 2b-10d (Table II). Some of the compounds gave poor melting points but all had correct elemental analyses (C, H, N) for the formulae listed in the table and NMR spectra consistent with the assigned structures.

Diethyl N-(4-Amino-2-fluorobenzoyl)-L-glutamate (28). Method C. A solution of the nitro compound 20 (20.0 g, 54 mmol) in EtOAc (400 mL) was stirred with 10% Pd-C (2.0 g) in an atmosphere of H_2 until complete reduction was indicated by TLC (ca. 4 h). The solution was filtered through Celite and evaporated to dryness to yield a white solid: 18.3 g (99%); mp 129-131 °C. Anal. (C₁₆H₂₁FN₂O₅) C, H, N, F.

The procedure was repeated with the appropriate nitro compounds 21, 23-27 to yield the amines 29, 31-35 (Table I).

Diethyl N-[2-Fluoro-4-(methylamino)benzoyl]-Lglutamate (36c). Method D. A mixture of 28 (1.0 g, 2.94 mmol) and MeI (0.55 mL, 8.82 mmol) in DMF (3 mL) was stirred for 1 h at 60 °C under argon. The cooled mixture was diluted with saturated aqueous NaHCO₃ and extracted with EtOAc. The dried EtOAc solution was evaporated and the residue was purified by chromatography using a gradient of 0–15% v/v EtOAc in CH₂Cl₂ as eluent. The product (321 mg, 31%) was isolated as a white solid: mp 97–98 °C; NMR (CDCl₃) δ 1.2, 1.3 (2 t, 6 H, 2 OCH₂CH₃), 2.0–2.55 (m, 4 H, CH₂CH₂CO₂Et), 2.9 (s, 3 H, NCH₃), 4.1, 4.2 (2 q, 4 H, 2 OCH₂CH₃), 4.85 (m, 1 H, CH), 6.2 (dd, 1 H, Ar 3-H), 6.4 (dd, 1 H, Ar 5-H), 7.1 (dd, 1 H, CONH), 7.9 (dd, 1 H, Ar 6-H). Anal. (C₁₇H₂₃FN₂O₅) C, H, N.

The procedure was repeated with the appropriate amines 28 and 29 using MeI, EtI, allyl bromide, and 2-bromoethyl acetate as the alkylating agents to yield 36d,e and 37c-e,g (Table I).

Diethyl N-[2-Fluoro-4-[(2-fluoroethyl)amino]benzoyl]-Lglutamate (36f). Method E. A mixture of 28 (6.0 g, 17.6 mmol), 1-bromo-2-fluoroethane (2.0 mL, 26.4 mmol), and 2,6-lutidine (2.05 mL, 17.6 mmol) in DMA (30 mL) was heated for 24 h at 120 °C in a sealed tube. The cooled reaction mixture was partitioned between EtOAc (3×70 mL) and H₂O (150 mL). The EtOAc solution was dried and the solvent was evaporated. The crude product was purified by chromatography using a gradient of 0-20% v/v EtOAc in CH₂Cl₂ as eluent. The product (5.67 g, 83%) was isolated as a gum: NMR (Me₂SO-d₆) δ 1.15, 1.20 (2 t, 6 H, 2 OCH₂CH₃), 2.0 (m, 2 H, CHCH₂CH₂CO₂Et), 2.4 (t, 2 H, CHCH₂CH₂CO₂Et), 3.4 (dq, 2 H, CH₂N), 4.05, 4.1 (2 q, 4 H, 2 OCH₂CH₃), 4.4 (m, 1 H, CH), 4.55 (dt, 2 H, CH₂F), 6.4 (dd, 1 H, Ar 3-H), 6.5 (dd, 1 H, Ar 5-H), 6.7 (t, 1 H, >NH), 7.5 (dd, 1 H, Ar 6-H), 7.85 (dd, 1 H, CONH). The procedure was repeated with the amine 29 to yield 37f (Table I).

Diethyl N-[4-[(2-Butyroxyethyl)amino]-2-fluorobenzoyl]-L-glutamate (36g). Method F. A mixture of 28 (6.1 g, 17.9 mmol), 2-bromoethyl butyrate (18.9 g, 9.68 mmol), NaI (2.68 g, 17.9 mmol), and 2,6-lutidine (4.17 mL, 35.8 mmol) in DMF (24 mL) was stirred for 3 h at 140 °C under argon. The cooled reaction mixture was partitioned between EtOAc $(2 \times 200 \text{ mL})$ and H_2O (80 mL) containing 5 M H_2SO_4 (5 mL). The EtOAc solution was washed repeatedly with H₂O until the washings had pH >6. The organic solution was dried and evaporated to dryness. The crude product was purified by chromatography using a gradient of 0-10% v/v EtOAc in CH₂Cl₂ as eluent. The product (4.07 g, 47%) was isolated as an oil: NMR (Me₂SO-d₆) $\delta 0.9$ (t, 3 H, CH₃), 1.15, 1.2 (2 t, 6 H, 2 OCH₂CH₃), 1.55 (m, 2 H, COCH₂CH₂CH₃), 2.0 (m, 2 H, CHCH₂CH₂CO₂Et), 2.3 (t, 2 H, CH₂CO₂), 2.35 (t, 2 H, CHCH₂CH₂CO₂Et), 3.4 (t, 2 H, CH₂N), 4.05, 4.1 (2 t, 4 H, 2 OCH₂CH₃), 4.15 (t, 2 H, OCH₂CH₂N), 4.45 (m, 1 H, CH), 6.4 (dd, 1 H, Ar 3-H), 6.5 (dd, 1 H, Ar 5-H), 7.5 (dd, 1 H, Ar 6-H), 7.85 (dd, 1 H, CONH).

Diethyl N-[4-[(3-Acetoxypropyl)amino]-2-fluorobenzoyl]-L-glutamate (36h). Method G. A mixture of 28 (2.0 g, 5.9 mmol), 1,3-dibromopropane (2.4 mL, 23.6 mmol), and 2,6-lutidine (0.76 mL, 6.5 mmol) in DMF (15 mL) was stirred for 5 h at 110 °C under argon. The reaction was worked up according to method F above to afford diethyl N-[4-[(3-bromopropyl)-amino]-2-fluorobenzoyl-L-glutamate (36l): 980 mg (36%); mp 60-61 °C. Anal. (C₁₉H₂₆BrFN₂O₅) C, H, N, Br, F.

A mixture of **361** (960 mg, 2.07 mmol) and powdered anhydrous NaOAc (1.7 g, 20.7 mmol) in DMF was stirred for 7.5 h at 80 °C under argon. The cooled reaction mixture was filtered and the filtrate evaporated to dryness. The crude product was purified by chromatography using 5% v/v EtOH in CH₂Cl₂ as eluent. The product **36h** (770 mg, 85%) was isolated as a gum: NMR (Me₂SO-d₆) δ 1.15, 1.2 (2 t, 6 H, 2 OCH₂CH₃), 1.85 (m, 2 H, CH₂CH₂CH₂), 2.0 (s, 3 H, OCOCH₃), 2.05 (m, 2 H, CHCH₂CH₂CO₂Et), 2.40 (t, 2 H, CHCH₂CH₂CO₂Et), 3.15 (t, 2 H, CH₂N), 4.0–4.2 (m, 6 H, 3 OCH₂), 4.45 (m, 1 H, CH), 6.35 (dd, 1 H, Ar 3-H), 6.45 (dd, 1 H, Ar 5-H), 6.55 (t, 1 H, >NH), 7.45 (dd, 1 H, Ar 6-H), 7.85 (t, 1 H, CONH). Anal. (C₂₁H₂₉FN₂O₇) C, H, N.

Di-tert -butyl-N-[2-Fluoro-4-[(2-phthalimidoethyl)amino]benzoyl]-L-glutamate (53). Method H. A mixture of di-tert-butyl N-(4-amino-2-fluorobenzoyl)-L-glutamate (52)⁵ (1.98 g, 5.0 mmol), 2-(bromoethyl)phthalimide (7.62 g, 30 mmol), NaI (4.5 g, 30 mmol), and 2,6-lutidine (3.5 mL, 30 mmol) in DMA (7.5 mL) was stirred for 5 h at 130 °C under argon. The reaction was worked up according to method F above to afford a gum: 1.10 g (39%); NMR (CDCl₃) δ 1.5 (s, 18 H, 2 t-Bu), 2.05 (m, 2 H, CHCH₂CH₂CO₂-t-Bu), 2.3 (t, 2 H, CHCH₂CH₂CO₂-t-Bu), 3.45 (br t, 2 H, CH₂NH), 4.0 (t, 2 H, CH₂NPhth), 4.4 (m, 1 H, CH), 6.25 (dd, 1 H, Ar 3-H), 6.4 (dd, 1 H, Ar 5-H), 7.2 (dd, 1 H, Ar 6-H), 7.65-7.9 (m, 5 H, phthalimide ArH's and CONH).

N-[4-[N-[(3,4-Dihydro-2-methyl-4-oxo-6-quinazolinyl)methyl]-N-(2-aminoethyl)amino]-2-fluorobenzoyl]-L-glutamic Acid (2i). Method I. The amino compound 53 was alkylated with 11 according to method A above to give 54 in 27% yield.

A mixture of 54 (360 mg, 0.49 mmol), 3-(dimethylamino)propylamine (0.37 mL, 2.92 mmol) and N,N-diisopropylethylamine (0.17 mL, 0.97 mmol) in MeOH (6 mL) was stirred for 6 h under reflux and argon. The resulting solution was evaporated to dryness and the crude product was purified by chromatography using a gradient of 0-20% v/v MeOH in CH₂Cl₂ containing 1% Et₃N to provide the di-*tert*-butyl ester (164 mg, 55%) of 2i as a gum. Hydrolysis of this ester at 60 °C according to method B yielded an amorphous solid: 50 mg (37%; 20% overall from 54); mp 210-215 °C (softened >190 °C); NMR (Me₂SO-d₆) δ 2.0 (m, 2 H, CHCH₂CH₂CO₂H), 2.25 (t, 2 H, CHCH₂CH₂CO₂H), 2.3 (s, 3 H, CH₃), 3.05 (m, 2 H, CH₂NH₂), 3.7 (m, 2 H, CH₂N<, 4.3 (m, 1 H, CH), 4.8 (br s, 2 H, ArCH₂N<), 6.7 (dd, 1 H, 3'-H), 6.95 (dd, 1 H, 5'-H), 7.5-7.7 (m, 3 H, 6'-H, quinazoline 8-H and 7-H), 7.85 (br s, 1 H, quinazoline 5-H), 7.9 (dd, 1 H, CONH); MS (FAB) m/z 498 [M - H]⁻. Anal. (C₂₄H₂₆FN₅O₆·1.25H₂O) C, H, N.

N-[4-[N-[(3,4-Dihydro-2-methyl-4-oxo-6-quinazolinyl)methyl]-N-(cyanomethyl)amino]-2-fluorobenzoyl]-L-glut-

Quinazoline Antifolate Thymidylate Synthase Inhibitors

amic Acid (2j). Method J. Alkylation of 52 with bromoacetonitrile at 95 °C using method F gave 55 (91%), which was further alkylated with 11 (method A) to afford 56 (20%) as a foam. Anal. $(C_{32}H_{38}FN_5O_6\cdot0.25H_2O)$ C, H, N.

A solution of 56 (615 mg, 1.01 mmol) in CF₃CO₂H (6.5 mL) was kept for 10 min and evaporated to dryness. The residue was dissolved in a solution of NaHCO₃ (840 mg, 10 mmol) in H₂O (10 mL). The solution was filtered into a centrifuge tube and brought to pH 3.0 with 2 N aqueous HCl. The precipitate was isolated by centrifugation and freed from inorganic ions by four cycles of aqueous suspension-centrifugation-decantation. The damp product was freeze-dried to give an off-white amorphous solid: 430 mg (84%); mp 160-162 °C (softened >145 °C); NMR (Me₂SO-d₆) δ 2.0 (m, 2 H, CHCH₂CH₂CO₂H), 2.3 (t, 2 H, CHCH₂CH₂CO₂H), 2.35 (s, 3 H, CH₃), 4.4 (m, 1 H, CH), 4.8 (br s, 2 H, CH₂CH₂CO), 4.85 (br s, 2 H, ArCH₂N<), 6.75 (dd, 1 H, 3'-H), 6.8 (dd, 1 H, 5'-H), 6.95-7.8 (m, 3 H, 6'-H, quinazoline 8-H and 7-H), 7.95 (d, 1 H, quinazoline 5-H), 8.0 (dd, 1 H, CONH); MS (FAB) m/z 494 [M - H]⁻. Anal. (C₂₄H₂₂FN₅O₆·0.5H₂O) C, H, N.

N-[4-[N-[(3,4-Dihydro-2-methyl-4-oxo-6-quinazolinyl)methyl]-N-(carbamoylmethyl)amino]-2-fluorobenzoyl]-Lglutamic Acid (2k). The diethyl ester 44k prepared as for 56 above from 28 was hydrolyzed according to method B. The product 2k (187 mg, 22%) had mp 185 °C (softened >170 °C); MS (FAB) m/z 512 [M - H]⁻. Anal. (C₂₄H₂₄FN₅O₇·H₂O) C, H, N.

Diethyl N-(3-Fluoro-4-nitrobenzoyl)-L-glutamate (21). Method K. A stirred slurry of 3-fluoro-4-nitrobenzoic acid⁵ (18.5 g, 0.1 mol) in dry toluene (400 mL) was treated with SOCl₂ (11 mL, 0.15 mol) and the mixture was heated under reflux for 1.5 h. The solution was cooled, filtered through Celite, and evaporated to dryness to give the crude acid chloride as a brown oil which solidified on standing. A solution of this acid chloride in CH₂Cl₂ (400 mL) was added over 15 min to a stirred mixture of diethyl L-glutamate hydrochloride (24 g, 0.1 mol) and Et₃N (27.9 mL, 0.2 mol) in CH₂Cl₂ (400 mL) below 25 °C (with cooling). The resulting brown solution was stirred for a further 2 h and then washed with H_2O (2 × 500 mL), treated with charcoal, and dried. The solution was evaporated to dryness and the residue was recrystallized from toluene-cyclohexane to give yellow needles: 35.75 g (96%); mp 80-81 °C; NMR (CDCl₃) δ 1.25, 1.30 (2 t, 6 H, 2 OCH₂CH₃), 2.25 (m, 2 H, CHCH₂CH₂CO₂Et), 2.5 (t, 2 H, CHCH₂CH₂CO₂Et), 4.15, 4.25 (2 q, 4 H, 2 OCH₂CH₃), 4.7 (m, 1 H, CH), 7.7 (d, 1 H, CONH), 7.75 (dd, 1 H, 6-H), 7.8 (dd, 1 H, 2-H), 8.15 (dd, 1 H, 5-H). Anal. $(C_{16}H_{19}FN_2O_7)$ C, H, N, F.

The procedure was repeated with 4-nitro-2-(trifluoromethyl)benzoic acid¹¹ to yield diethyl N-[4-nitro-2-(trifluoromethyl)benzoyl]-L-glutamate (23).

Diethyl N-[3-Fluoro-4-(prop-2-ynylamino)benzoyl]-Lglutamate (37a). Method L. A mixture of 29 (6.8 g, 20 mmol) (prepared from 21 according to method C), K₂CO₃ (2.76 g, 20 mmol), and propargyl bromide (4.46 mL of an 80% w/w solution in toluene, 40 mmol) in DMF (22 mL) was stirred for 3 h at 120 °C under argon. The cooled mixture was filtered and the filtrate was evaporated to dryness. The residue was partitioned between EtOAc (250 mL) and H₂O (250 mL). The organic phase was dried and the solvent was evaporated. The crude product was purified by gas chromatography using 40% v/v EtOAc in hexane as eluent. The product (4.02 g, 53%) was isolated as a yellow oil: NMR (CDCl₃) § 1.25, 1.3 (2 t, 6 H, 2 OCH₂CH₃), 2.2 (m, 2 H, CHCH₂CH₂CO₂Et), 2.25 (t, 1 H, C=CH), 2.45 (t, 2 H, CHCH₂CH₂CO₂Et), 4.0 (br s, 2 H, CH₂C=C), 4.1, 4.45 (2 q, 4 H, 2 OCH₂CH₃), 4.55 (br s, 1 H, HN<), 4.75 (m, 1 H, CH), 6.8 (t, 1 H, 5-H), 6.9 (d, 1 H, CONH), 7.5 (dd, 1 H, 6-H), 7.55 (dd, 1 H, 2-H). Anal. (C₁₉H₂₃FN₂O₅) H, N, F; C: calcd, 59.6; found, 60.3.

Diethyl N-[4-[(3-Acetoxypropyl)amino]-3-fluorobenzoyl]-L-glutamate (37 h). Method M. A mixture of 29 (340 mg, 1.0 mmol), 3-bromopropyl acetate (1.09 g, 6 mmol), and 2,6-lutidine (0.13 mL, 1.1 mmol) in DMF (3 mL) was stirred for 9 h at 150 °C under argon. The cooled mixture was partitioned between EtOAc (50 mL) and 0.5 N aqueous H₂SO₄ (10 mL). The organic solution was washed several times with brine until the washings had pH >6.0, dried, and concentrated to a gum. Purification was achieved by chromatography using a gradient of 0–10% v/v EtOAc in CH₂Cl₂ as eluent. The product (117 mg, 27%) was isolated as a gum: NMR (Me₂SO-d₆) δ 1.15 (2 t, 6 H, 2 OCH₂CH₃), 1.9 (m, 2 H, CH₂CH₂CH₂), 2.0 (s, 3 H, OCOCH₃), 2.05 (m, 2 H, CHCH₂CO₂Et), 2.4 (t, 2 H, CHCH₂CO₂Et), 3.2 (t, 3 H, NCH₂), 4.1 (m, 6 H, 3 OCH₂), 4.4 (m, 1 H, CH), 6.05 (br s, 1 H, amine NH), 6.7 (dd, 1 H, Ar 5-H), 7.55 (d, 1 H, Ar 6-H), 7.6 (d, 1 H, Ar 2-H), 8.3 (d, 1 H, CONH).

The procedure was repeated with the appropriate (4-aminobenzoyl)-L-glutamate esters and alkyl halides at the stated temperatures (Table I) to yield the amines 38d and 40c-43d.

Diethyl N-[4-[N-[[3,4-Dihydro-2-methyl-4-oxo-3-[(pivaloyloxy)methyl]-6-quinazolinyl]methyl]-N-(3-acetoxypropyl)amino]-3-fluorobenzoyl]-L-glutamate (45 h). Method N. A mixture of 37h (110 mg, 0.25 mmol), the bromomethyl compound 577 (183 mg, 0.5 mmol), and 2,6-lutidine (0.06 mL, 0.5 mmol) in DMF (2 mL) was stirred for 7.5 h at 105 °C under argon. The cooled mixture was evaporated to dryness and the residue was purified by chromatography using a gradient of 0-30% v/vEtOAc in CH_2Cl_2 as eluent. The product (102 mg, 56%) was isolated as a gum: NMR (Me₂SO- d_6) δ 1.15 (s, 9 H, t-Bu), 1.2 (2 t, 6 H, 2 OCH₂CH₃), 1.9 (m, 2 H, CH₂CH₂CH₂), 1.95 (s, 3 H, OCOCH₃), 2.05 (m, 2 H, CHCH₂CH₂CO₂Et), 2.4 (t, 2 H, CHCH₂CH₂CO₂Et), 2.6 (s, 3 H, CH₃), 3.35 (m, 2 H, CH₂N<), 4.05 $(m, 6 H, 3 OCH_2), 4.4 (m, 1 H, CH), 4.6 (br s, 2 H, ArCH_2N<),$ 6.05 (s, 2 H, OCH₂N), 7.05 (dd, 1 H, 5'-H), 7.55 (d, 1 H, 6'-H), 7.6 (d, 1 H, 2'-H), 7.7 (m, 2 H, quinazoline 7-H and 8-H), 8.05 (d, 1 H, quinazoline 5-H), 8.5 (d, 1 H, CONH).

The procedure was repeated with the appropriate starting amines to yield the 3-[(pivaloyloxy)methyl]quinazoline diesters **45e** and **45g** (Table I).

Diethyl N-(2-Methoxy-4-nitrobenzoyl)-L-glutamate (27). Method O. Oxalyl chloride (2.21 mL, 25.3 mmol) was added dropwise over 30 min to a stirred suspension of 2-methoxy-4nitrobenzoic acid¹² (5.0 g, 25.3 mmol) in CH₂Cl₂ (200 mL) containing DMF (1 drop). After 3 h the solvent was evaporated and the crude acid chloride was dissolved with stirring in a mixture of CH₂Cl₂ (200 mL) and Et₃N (17.5 mL) and cooled to 0 °C. A solution of diethyl L-glutamate hydrochloride (6.08 g, 25.3 mmol) in CH_2Cl_2 (50 mL) was added dropwise while the temperature was kept below 20 °C. Stirring was continued for 1 h and the mixture was washed with 2 N HCl $(2 \times 50 \text{ mL})$ and brine (50 mL). The organic phase was dried and evaporated to dryness. The product (8.98 g, 93%) had the following: NMR (CDCl₃) δ 1.2, 1.35 (2 t, 6 H, 2 OCH₂CH₃), 2.15 (m, 2 H, CHCH₂CH₂CO₂Et), 2.45 (t, 2 H, CHCH₂CH₂CO₂Et), 4.1, 4.25 (2 q, 4 H, 2 OCH₂CH₃), 4.35 (s, 3 H, OCH₂), 4.85 (m, 1 H, CH), 7.85 (d, 1 H, Ar 3-H), 7.9 (dd, 1 H, Ar 5-H), 8.35 (d, 1 H, Ar 6-H), 8.5 (d, 1 H, CONH).

The procedure was repeated with the substituted 4-nitrobenzoic acids 17^{13} and 18^{14} to yield the substituted (4-nitrobenzoyl)-glutamate esters 25 and 26.

Diethyl N-[4-[N-[(3,4-Dihydro-2-methyl-4-oxo-6quinazolinyl)methyl]-N-prop-2-ynylamino]-2-methoxybenzoyl]-L-glutamate (51a). Method P. A mixture of 11 (1.0 g, 3.95 mmol), 43a (1.54 g, 3.95 mmol) (prepared from 27 according to methods C and M), and 2,6-lutidine (2.3 mL, 19.7 mmol) in DMF (10 mL) was stirred for 18 h at 80 °C under argon. The cooled mixture was partitioned between EtOAc (4 ×:50 mL) and H₂O (100 mL). The combined organic phase was washed with brine, dried, and evaporated to dryness. The crude oil was purified by chromatography eluting with 5% v/v MeOH in EtOAc. The product (780 mg, 35%) was isolated as a gum: NMR (Me₂SO-d₆) δ 1.15, 1.2 (2 t, 6 H, 2 OCH₂CH₃), 2.0 (m, 2 H, CHCH₂CH₂CO₂Et), 2.3 (t, 2 H, CHCH₂CH₂CO₂Et), 2.35 (s, 3 H, CH₃), 3.25 (t, 1 H, C==CH), 3.85 (s, 3 H, OCH₃), 4.1, 4.2 (2 q, 4 H, 2 OCH₂CH₃), 4.35 (d, 2 H, CH₂C==C), 4.45 (m, 1 H, CH), 4.8 (br s, 2 H), ArCH₂N<), 6.45 (s, 1 H, 3'-H), 6.6 (d, 1 H, 5'-H), 7.55 (d, 1 H, quinazoline

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8-H), 7.7 (d and dd, 2 H, 6'-H and quinazoline 7-H), 8.0 (d, 1 H, quinazoline 5-H), 8.2 (d, 1 H, CONH).

The procedure was repeated with the appropriate amines to yield the quinazoline antifolate diesters 45f, 46a-51d (Table I).

Methyl 2-Nitro-4-(prop-2-ynylamino)benzoate (59). A mixture of methyl 4-amino-2-nitrobenzoate (58)⁸ (10.6 g, 54 mmol), 2,6-lutidine (8.14 mL, 70 mmol), and propargyl bromide (7.82 mL of an 80% w/w solution in toluene, 70 mmol) in DMA (50 mL) was stirred for 4 h at 80 °C under argon. A second portion of propargyl bromide (70 mmol) was added and stirring was continued for a further 4.5 h at 80 °C. The cooled reaction mixture was partitioned between EtOAc (300 mL) and H₂O (3×100 mL). The organic phase was dried and evaporated to dryness. The crude product was purified by chromatography using CH₂Cl₂ as eluent to yield a buff solid: 6.58 g (52%); mp 134-135 °C. Anal. (C₁₁H₁₀N₂O₄) C, H; N: calcd, 12.0; found, 11.5.

Met hyl 4-[N-[(3,4-Dihydro-2-met hyl-4-oxo-6quinazolinyl)methyl]-N-prop-2-ynylamino]-2-nitrobenzoate (60). A mixture of the amino ester 59 (2.0 g, 8.55 mmol), the bromomethyl compound 11 (2.6 g, 10.25 mmol), and 2,6-lutidine (2.0 mL, 17.1 mmol) in DMA (10 mL) was stirred for 4 h at 80 °C under argon. The cooled reaction mixture was partitioned between EtOAc (2 × 300 mL) and H₂O (150 mL). The organic phase was washed with H₂O, dried, and evaporated to dryness. Trituration of the residue with EtOAc gave an amorphous buff solid: 1.52 g (44%); NMR (Me₂SO-d₆) δ 2.35 (s, 3 H, CH₃), 3.3 (t, 1 H, C=CH), 3.75 (s, 3 H, OCH₃), 4.45 (d, 2 H, CH₂C=C), 4.9 (br s, 2 H, ArCH₂N<), 7.05 (dd, 1 H, 5'-H), 7.2 (d, 1 H, 3'-H), 7.55 (d, 1 H, quinazoline 8-H), 7.65 (dd, 1 H, quinazoline 7-H), 7.75 (d, 1 H, 6'-H), 7.95 (d, 1 H, quinazoline 5-H).

4-[N-[(3,4-Dihydro-2-methyl-4-oxo-6-quinazolinyl)methyl]-N-prop-2-ynylamino]-2-nitrobenzoic Acid (61). The methyl ester 60 (1.50 g, 3.7 mmol) was stirred for 2 h under argon in a mixture of 1 N aqueous NaOH (18.5 mL, 18.5 mmol) and EtOH (18.5 mL). The resulting solution was evaporated below 30 °C to ca. 10 mL, filtered, and acidified to pH 1 with 2 N aqueous HCl. The precipitated buff solid was filtered off, washed with H₂O, and vacuum dried: 1.24 g (82%); mp 260–262 °C (dec). Anal. (C₂₀H₁₆N₄O₅·H₂O) C, H, N.

Diethyl N-[4-[N-[(3,4-Dihydro-2-methyl-4-oxo-6quinazolinyl)methyl]-N-prop-2-ynylamino]-2-nitrobenzoyl]-L-glutamate (62a). Method Q. A solution of the acid 61 monohydrate (392 mg, 0.96 mmol) and diethyl L-glutamate hydrochloride (720 mg, 3.0 mmol) in DMF (3.3 mL) was stirred at 0 °C during the dropwise addition over 15 min of diphenyl phosphorazidate (0.43 mL, 2.0 mmol) followed by Et₃N (0.98 mL, 7.0 mmol) again over 15 min. The reaction mixture was allowed to warm to room temperature overnight and then partitioned between CH₂Cl₂ (3 × 25 mL) and H₂O (25 mL). The organic phase was washed with H₂O, dried, and evaporated to dryness. The crude product was purified by chromatography using a gradient of 0–6% v/v EtOH in CH₂Cl₂ as eluent to give a buff solid: 421 mg (73%); mp 158-163 °C; NMR (Me₂SO-d₆) δ 1.15, 1.2 (2 t, 6 H, 2 OCH₂CH₃), 2.0 (m, 2 H, CHCH₂CH₂CO₂Et), 2.35 (s, 3 H, CH₃), 2.4 (t, 2 H, CHCH₂CH₂CO₂Et), 3.25 (t, 1 H, C=CH), 4.05, 4.1 (2 q, 4 H, 2 OCH₂CH₃), 4.35 (m, 1 H, CH), 4.45 (d, 2 H, CH₂C=C), 4.85 (br s, 2 H, ArCH₂N<), 7.1 (dd, 1 H, 5'-H), 7.25 (d, 1 H, 3'-H), 7.45 (d, 1 H, 6'-H), 7.55 (d, 1 H, quinazoline 8-H), 7.7 (dd, 1 H, quinazoline 7-H), 7.95 (d, 1 H, quinazoline 5-H), 8.8 (br d, CONH). Anal. (C₂₉H₃₁N₅O₈·0.75H₂O) C, H; N: calcd, 11.85; found, 11.3.

Registry No. 2a, 112887-67-9; 2b, 112887-95-3; 2c, 112887-65-7; 2d, 112887-66-8; 2e, 129175-19-5; 2f, 112887-97-5; 2g, 112887-96-4; 2h. 129175-20-8; 2i, 129175-21-9; 2j, 129175-22-0; 2k, 112887-98-6; 3a, 112888-30-9; 3b, 129175-23-1; 3c, 129175-24-2; 3d, 129175-25-3; 3e. 129175-26-4; 3f. 129175-27-5; 3g, 129175-28-6; 3h, 129175-29-7; 4a, 112888-33-2; 4d, 112888-32-1; 5a, 112888-26-3; 6a, 129175-30-0; 6c, 129175-31-1; 6d, 129175-32-2; 7a, 129193-64-2; 7c, 129175-33-3; 7d, 129175-34-4; 8a, 112888-31-0; 9a, 112888-28-5; 9d, 112888-27-4; 10a, 129175-35-5; 10d, 112888-34-3; 11, 112888-43-4; 13, 403-21-4; 15, 320-37-6; 17, 91533-09-4; 18, 17336-10-6; 19, 2597-56-0; 20, 106585-55-1; 21, 126632-34-6; 23, 112888-88-7; 24, 80015-10-7; 25, 129175-84-4; 26, 129175-85-5; 27, 129175-86-6; 28, 106585-56-2; 29, 126632-35-7; 30, 80014-92-2; 31, 112888-89-8; 32, 80014-85-3; 33, 129175-80-0; 34, 129175-81-1; 35, 129175-82-2; 36a, 106585-57-3; 36c, 112888-51-4; 36d, 129175-62-8; 36e, 129175-63-9; 36f, 129175-64-0; 36g, 129175-65-1; 36h, 129175-66-2; 36k, 112888-53-6; 361, 129175-83-3; 37a, 1128893-4; 37c, 129175-69-5; 37d, 129175-70-8; 37e, 129175-71-9; 37f, 129175-72-0; 37g, 129175-73-1; 37h, 129175-74-2; 38a, 80014-87-5; 38d, 112888-95-6; 39a, 112888-86-5; 40a, 80014-86-4; 40c, 129175-75-3; 40d, 129175-76-4; 41a, 129175-77-5; 41c, 129175-78-6; 41d, 129175-79-7; 42a, 112888-94-5; 43a, 112888-91-2; 43d, 112888-90-1; 44a, 129175-36-6; 44b, 129175-37-7; 44c, 129175-38-8; 44d, 129175-39-9; 44e, 129175-40-2; 44f, 129193-65-3; 44g, 129175-41-3; 44h, 129175-42-4; 44k, 129193-66-4; 45a, 129193-67-5; 45b, 129175-43-5; 45c, 129175-44-6; 45d, 129175-45-7; 45e, 129193-68-6; 45f, 129193-69-7; 45g, 129175-46-8; 45h, 129175-47-9; 46a, 129175-48-0; 46d, 129175-49-1; 47a, 112888-87-6; 48a, 129175-50-4; 48c, 129175-51-5; 48d, 129175-52-6; 49a, 129175-53-7; 49c, 129175-54-8; 49d, 129175-55-9; 50a, 129175-56-0; 51a, 129175-57-1; 51d, 129175-58-2; 52, 85803-27-6; 53, 129175-67-3; 54, 129175-60-6; 55, 129175-68-4; 56, 129175-61-7; 57, 112888-39-8; 58, 84228-45-5; 59, 129175-87-7; 60, 129175-88-8; 61, 129175-89-9; 62a, 129175-59-3; 62d, 112888-98-9; TS, 9031-61-2; BrCH₂C=CH, 106-96-7; BrCH₂CH=CH₂, 106-95-6; BrCH₂CH₂OAc, 927-68-4; BrCH₂CH₂F, 762-49-2; BrCH₂CH₂OC-OCH₂CH₂CH₃, 6065-66-3; Br(CH₂)₃Br, 109-64-8; Br(CH₂)₃OAc, 592-33-6; H-Glu(OEt)-OEt·HCl, 1118-89-4; 2-(bromoethyl)phthalimide, 574-98-1.